



Characterization of rare-earth aluminum oxynitrides synthesized by carboreduction and nitridation

François Cheviré, Arthur Pallu, Franck Tessier

► To cite this version:

François Cheviré, Arthur Pallu, Franck Tessier. Characterization of rare-earth aluminum oxynitrides synthesized by carboreduction and nitridation. ISNT 2009 - 6th International Conference on Nitride and Related Materials, Mar 2009, Karlsruhe, Germany. hal-01261259

HAL Id: hal-01261259

<https://hal.science/hal-01261259>

Submitted on 25 Jan 2016

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Characterization of Rare-Earth Aluminum Oxynitrides Synthesized by Carbo-reduction and Nitridation



François Cheviré, Arthur Pallu, Franck Tessier

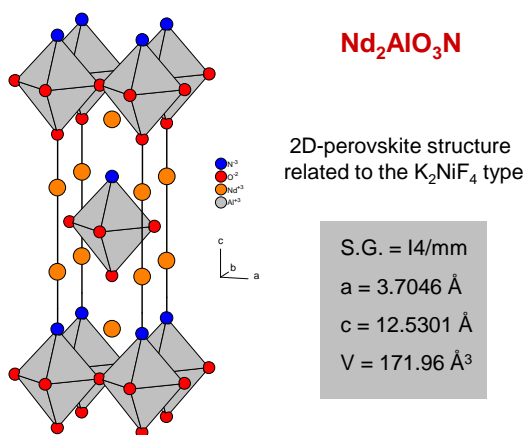


UMR CNRS 6226 "Sciences Chimiques de Rennes" – Equipe Verres et Céramiques
Groupe Matériaux Nitrures, 35042 – 35042 Rennes Cedex – France

Background

Over the recent years, a great interest has been shown for multinary oxynitrides containing main group elements especially in the fields of phosphors or visible-light driven photocatalysts. However if a tremendous work has been done on nitridosilicates and nitridoaluminosilicates based materials, very little is known about multinary (oxy)nitride systems based on aluminum (R_2AlO_3N , $Al_{11}O_{12}N$), gallium (La_2GaN_3) or phosphorous (Zn_2PO_3N). A large part of our research is then devoted to the preparation of new materials among those systems by exploring less-common synthesis routes than the usual solid state reaction or the ammonia flow-based nitridation. As a first step of the present work, we have successfully reinvestigated the preparation of the R_2AlO_3N oxynitrides ($R = Nd$ and Sm) using the carbo-reduction and nitridation process (CRN).

Structure...



R. Marchand et al. – Rev. Chim. Minér. 19 (1982) p684

Synthesis routes...

Previous synthesis : mixture of Nd_2O_3 or Sm_2O_3 and AlN sealed under N_2 in nickel tubing and fired at $1350^\circ C$ with repeated heating.

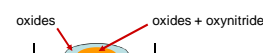
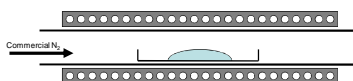
Long reaction process (~ 6 x 24h)

Very difficult to obtain pure samples $\rightarrow RAlO_3$ and R_2O_3 minor phases

Carbo-reduction & nitridation (CRN):

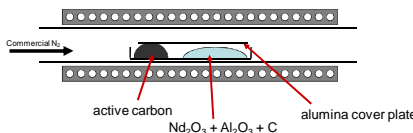


Regular apparatus



\rightarrow reoxidation during synthesis

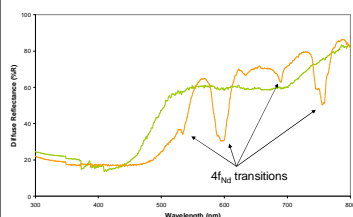
"Oxygen trap" apparatus



\rightarrow oxides formation efficiently reduced

Optical properties...

$Nd_2AlO_3N \rightarrow$ orange powder
 $Sm_2AlO_3N \rightarrow$ greenish yellow powder

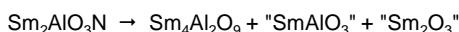


$\lambda_{Sm} \sim 480 \text{ nm}$
 \Rightarrow Yellow

$\lambda_{Nd} \sim 530 \text{ nm}$
 \Rightarrow Orange

Thermal analyses...

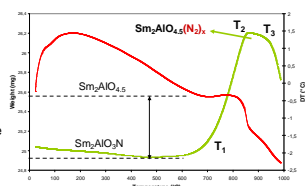
Oxidation under air starts $\sim 600^\circ C$



$T_1 \rightarrow T_2$: Oxidation with N_2 retention

$T_2 \rightarrow T_3$: "Intermediate phase"

$T_3 \rightarrow$: Oxide formation with N_2 release

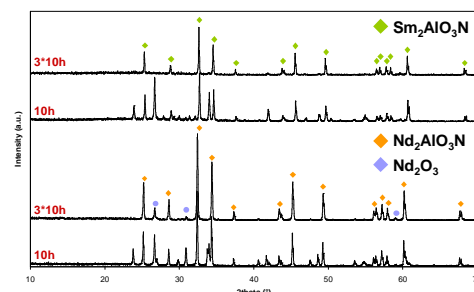


X-Ray Diffraction...

$\Rightarrow Sm_2AlO_3N$ can be obtained as a nearly pure sample after only 3 heating at $1250^\circ C$

\Rightarrow In the same conditions, a Nd_2O_3 impurity phase still be present along with Nd_2AlO_3N

little oxidation remains



Conclusion...

\triangleright Carbo-reduction & Nitridation (CRN) is a convenient way to prepare aluminum based oxynitrides with interesting optical properties and good thermal stability under air.

\triangleright Apparatus will be upgraded with a nitrogen gas purification system (moisture and oxygen traps, hot titanium wire) to avoid reoxidation of the samples.

\triangleright Other systems are under investigation such as $A_{2-x}R_xAlO_{4-x}N_x$ solid solution with A = Alkaline earth and R = Rare earth.

Contact informations:

francois.chevire@univ-rennes1.fr

+33 (0) 2 23 23 62 34

http://www.verceram.univ-rennes1.fr

Fax +33 (0) 2 23 23 56 83